

# Selective conjugate addition to zerumbone and transannular cyclization of its derivatives

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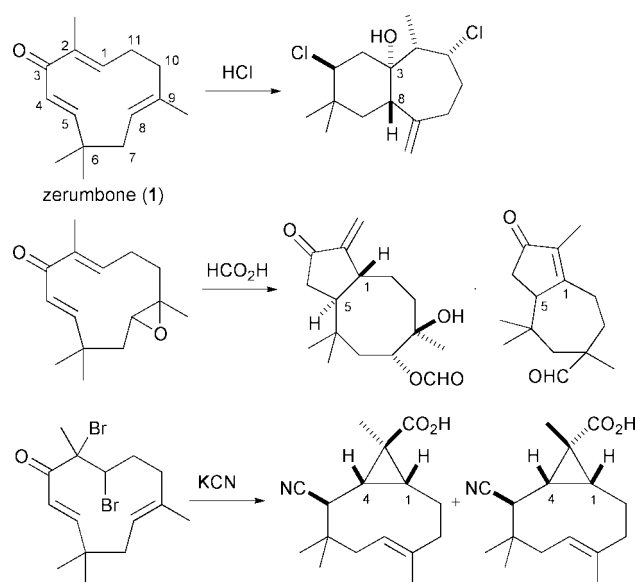
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Some chemical transformations of zerumbone (**1**) isolated from rhizomes of *Zingiber zerumbet* Smith have been described. Regioselective addition of benzenethiol and benzeneselenol to the  $\Delta^{1,2}$  double bond of **1** occurs in the presence of tetra-*n*-butylammonium fluoride (TBAF) as a catalyst to give 1-phenylthio and 1-phenylseleno adducts **2a** and **2b**, respectively, with moderate diastereoselectivity. In the presence of  $\text{BF}_3 \cdot \text{OEt}_2$ , the reaction of **2a** and **2b** with trimethylsilyl cyanide proceeds in 1,4-addition fashion to give the corresponding 5-cyano silyl enolates **3a** and **3b**, respectively. Desilylation of **3a** and **3b** with  $\text{HF} \cdot \text{pyridine}$  provides (8*E*)-5-cyano-2,6,6,9-tetramethyl-1-phenylthiocycloundec-8-en-3-one (**5a**) and (8*E*)-5-cyano-2,6,6,9-tetramethyl-1-phenylselenocycloundec-8-en-3-one (**5b**) as a single diastereoisomer, respectively. Treatment of **2a** and **2b** with  $\text{TiCl}_4$  affords compounds with a bicyclo[5.3.0]decane skeleton, **6a** and **6b**, respectively, as *trans*-fused isodaucane derivatives *via* a transannular reaction between C-4 and C-9 of **2** followed by rearrangement. Palladium-mediated transannular reaction between C-4 and C-9 of **3b** yields **7b** with a bicyclo[5.4.0]undec-8-ene skeleton. Brønsted acid-catalyzed transannular reactions between C-3 and C-8 of **5a** and **5b** afford compounds with bicyclo[5.4.0]undecane skeletons **8a** and **8b**. A lithium enolate of (1*Z*)-5-cyano-8,9-epoxyzerumbone **10/10'** derived from **5b** gives compounds with tricyclic skeletons, **11** and **12**, by a transannular reaction between C-4 and C-8. Compounds **11** and **12** involve the carbon skeleton of 5,8-fused bicyclic natural sesquiterpene, asteriscane. Some molecular structures of key intermediates and products have been characterized by X-ray crystallography.

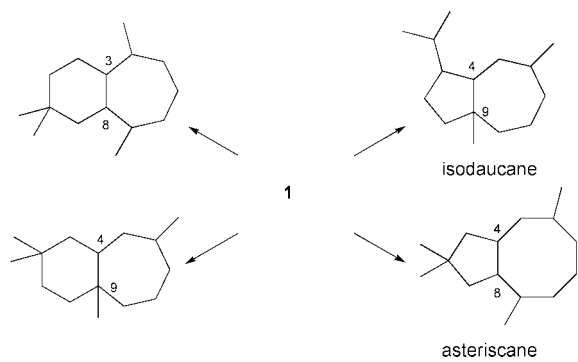
## Introduction

Transannular cyclizations are of much importance in the synthesis of polycyclic complex molecules.<sup>1</sup> They occur frequently in medium, *i.e.* 8-, 9-, 10-, and 11-membered, rings where favored conformations allow opposite sides of the rings to come close to each other for bond formation. Transannular cyclizations involving carbocations generated from cyclodeca- and cycloundeca-polyenes have been the subject of extensive research, largely in connection with our understanding of the biosynthesis of a wide range of polycyclic sesquiterpenes.<sup>2</sup> Among the 10- and 11-membered sesquiterpenes, germacrene<sup>3</sup> and humulene<sup>4</sup> are known as the key precursors to eudesmane, guaiane, eremophilane, pseudoguianane, caryophyllane, protoilludane, hirustane, capnellane, and pentalenane *etc.*, and extensive studies on their transannular reactivity have demonstrated the suitability for transformation into such natural products. On the other hand, annulation of zerumbone (**1**)<sup>5</sup> as the 3-keto form of humulene, which has a unique cross-conjugated dienone unit, has been restricted to a few examples as shown in Scheme 1. Chhabra and co-workers reported the acid-catalyzed transannular cyclization between C-3 and C-8 of **1** leading to bicyclo[5.4.0]undecane skeletons.<sup>6</sup> Luu *et al.* reported that the Nazarov-type cyclization between C-1 and C-5 of zerumbone 8,9-epoxide, one component of *Zingiber zerumbet*, afforded bicyclo[6.3.0]undecane and bicyclo[5.3.0]decane skeletons.<sup>7</sup> Recently, we have reported the formation of bicyclo[7.1.0]decane skeletons *via* Favorskii-like ring-contractive transannular reaction between C-1 and C-4 of the 1,2-dibromozerumbone derivative.<sup>8</sup> In line with our continuing study into transformations of **1** we wish to report



Scheme 1

herein reactions of **1** involving new types of transannular cyclizations which provide several polycyclic compounds such as 6,7-, 5,7-, and 5,8-carbocycles, as illustrated in Scheme 2. The 5,7- and 5,8-bicyclic products involve the representative structural unit of natural sesquiterpenes, *i.e.*, isodaucane<sup>9</sup> and asteriscane.<sup>10</sup>

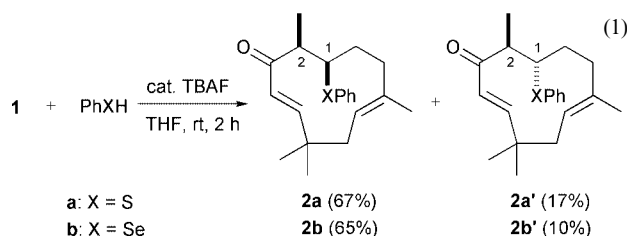


Scheme 2

## Results and discussion

### Introduction of functional groups into zerumbone. Conjugate addition reactions

As previously reported, conjugate addition of  $\text{CN}^-$  or  $\text{MeO}^-$  to zerumbone (**1**) is facile.<sup>8</sup> Other conjugate additions to **1** have since been examined. Thus, when the reaction of **1** with benzenethiol was carried out in tetrahydrofuran (THF) at room temperature in the presence of tetra-*n*-butylammonium fluoride (TBAF),<sup>11</sup> addition took place at C-1 to give a mixture of (4*E*,8*E*)-2,6,6,9-tetramethyl-1-phenylthiocycloundeca-4,8-dien-3-one (**2a**) and its diastereoisomer **2a'** in 67 and 17% isolated yields, respectively [eqn. (1)]. The relative stereo-



chemistry at the C-1 and C-2 centers was deduced by X-ray crystallography of a derivative of **2a** (*vide infra*). The reaction with benzeneselenol under the same reaction conditions gave a mixture of (4*E*,8*E*)-2,6,6,9-tetramethyl-1-phenylselenocycloundeca-4,8-dien-3-one (**2b**) and its diastereoisomer **2b'** in 65 and 10% yields, respectively. No adducts derived from addition to C-5 were observed in these reactions. Next, we examined Michael-type additions of organomagnesium reagents to **2**. The reaction of **2b** with ethylmagnesium bromide in the presence of  $\text{CuBr}\cdot\text{SMe}_2$  and trimethylsilyl chloride<sup>12</sup> afforded (8*E*)-5-ethyl-2,6,6,9-tetramethyl-1-phenylselenocycloundec-8-en-3-one as a single diastereoisomer in 42% yield after hydrolysis. However, the reaction of **2b** with other organomagnesium reagents such as trimethylsilylethynylmagnesium bromide and 2-(1,3-dioxolan-2-yl)ethylmagnesium bromide formed no adducts at C-5 under the same reaction conditions. Next we examined the addition of trimethylsilyl cyanide (TMSCN) as a carbon nucleophile. In the event, the conjugate addition of TMSCN to the remaining conjugated bond of **2b** was successfully achieved in the presence of 10 mol% of  $\text{BF}_3\cdot\text{OEt}_2$ <sup>13</sup> under reflux in  $\text{CHCl}_3$ , providing 5-cyano-2,6,6,9-tetramethyl-1-phenylseleno-3-(trimethylsilyloxy)cycloundeca-3,8-diene (**3b**) in 64% yield as a 10:1 mixture of (*E*)- and (*Z*)-silyl enolates [eqn. (2)]. The introduction of other functionalities at C-5 of **2** was not successful.

### Stereochemistry of **2**

When **3a**, obtained from the reaction of **2a** with TMSCN, was treated with aqueous HF in MeCN at room temperature for 4 h, (8*E*)-5-cyano-2,6,6,9-tetramethyl-1-phenylthiocycloundec-

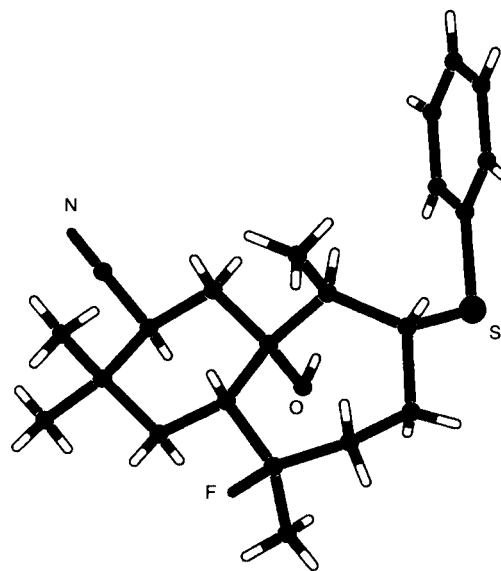
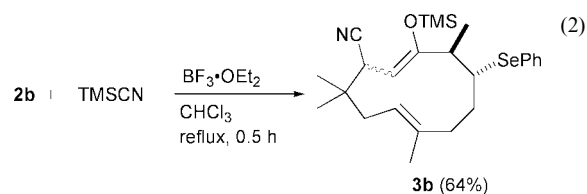
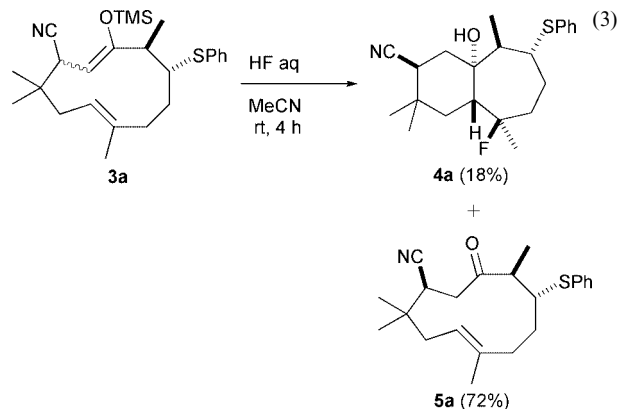


Fig. 1 Molecular structure of **4a** as determined by X-ray diffraction.



8-en-3-one (**5a**) was obtained as a major product (72% yield) together with a compound having a bicyclo[5.4.0]undecane skeleton **4a** in 18% yield [eqn. (3)]. The structure of **4a** was



unambiguously established by X-ray diffraction analysis. The X-ray structure indicates that **4a** has an axial methyl at C-2 and an equatorial phenylthio group at C-1 of a seven-membered carbocycle as shown in Fig. 1. A *trans* stereochemistry at ring junctures is also clarified by the X-ray analysis. The configurations of all the stereogenic centers except for the C-1 carbon are consistent with the stereochemistry of a similar bicyclo[6.3.0]undecane skeleton reported in ref. 6. Hence, the *syn*-1,2-stereochemical relationship at C-1 and C-2 of **2a** and **2b** is correct as illustrated in eqn. (1).

### Transannular cyclization of **2** followed by rearrangement leading to the compounds having a 5,7-bicyclic skeleton

Parker *et al.* have reported that the treatment of humulene with aqueous sulfuric acid in acetone afforded a bicyclo[5.3.0]deca-1,8-diene, *via* the initial formation of humulol (Scheme 3).<sup>14</sup> This stimulated us to investigate the acid-promoted transannular cyclization of zerumbone derivatives. When the reaction of **2a** with HCl gas was carried out in  $\text{CH}_2\text{Cl}_2$

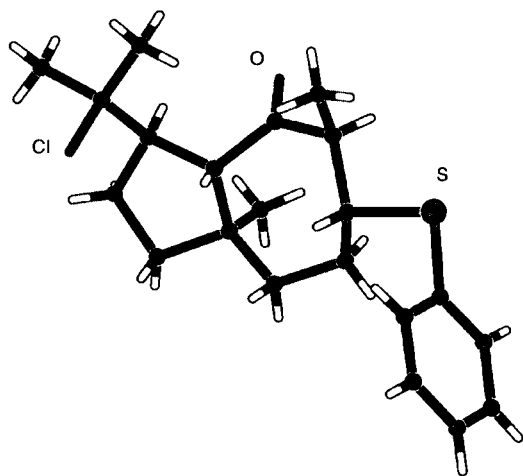
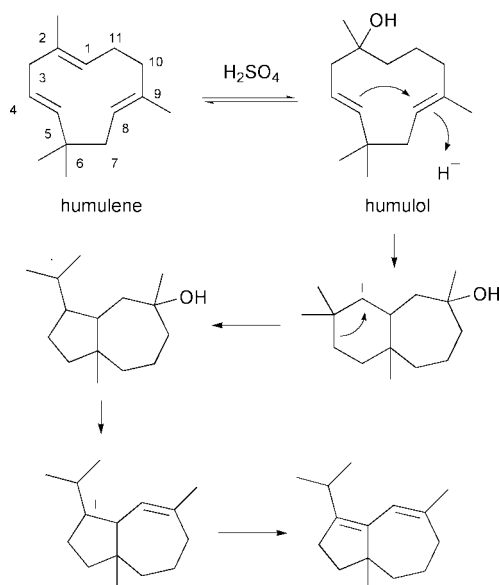
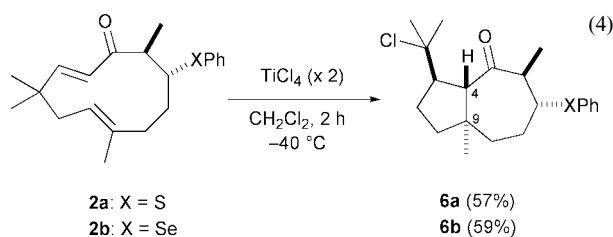


Fig. 2 Molecular structure of **6a**.

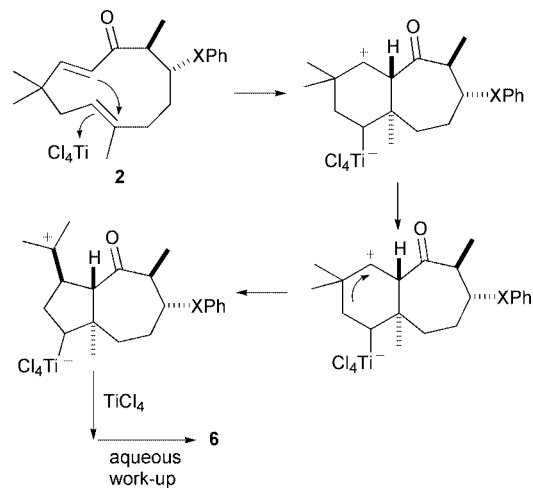


Scheme 3

at  $-78\text{ }^{\circ}\text{C}$ , a complex mixture was obtained, its IR spectra indicating the absence of the carbonyl group. We supposed that transannular cyclization similarly gave 6,7-bicyclic skeletons as shown in eqn. (3).<sup>15</sup> Next, we attempted the reaction of **2** with a Lewis acid. The reaction of **2a** with two equivalents of  $\text{TiCl}_4$  in  $\text{CH}_2\text{Cl}_2$  at  $-40\text{ }^{\circ}\text{C}$  afforded a bicyclo[5.3.0]decane skeleton **6a** in 57% yield as shown in eqn. (4). The structure of **6a** was determined by X-ray diffrac-



tion analysis (Fig. 2). The X-ray structure shows that the 5,7-bicyclic skeleton is formed *via* transannular cyclization between C-4 and C-9 followed by rearrangement. The structure of **6a** is similar to the isodaucane sesquiterpenes except for *trans* angular stereochemistry. The most plausible pathway to **6a** is shown in Scheme 4. The proposed initial coordination of  $\text{TiCl}_4$  with the  $\Delta^{8,9}$  double bond generates a cationic site at C-9 which then acts as the trigger for the subsequent cyclization

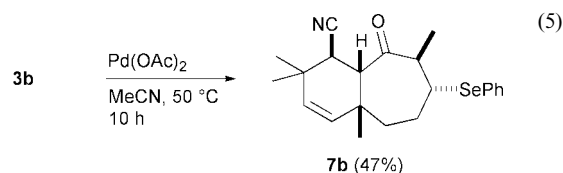


Scheme 4

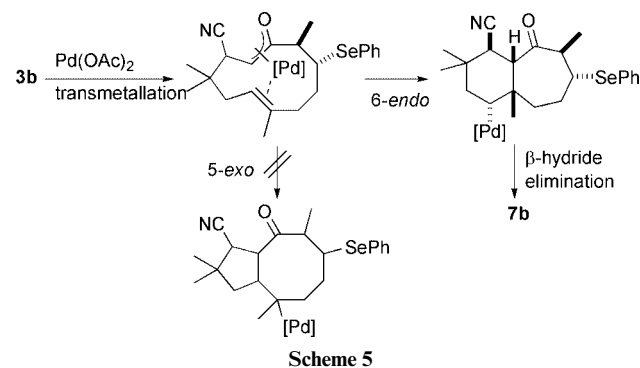
step involving participation by the  $\Delta^{4,5}$  double bond to provide a 6,7-bicyclic skeleton. The subsequent migration of an adjacent bond to the cationic center in the 6,7-bicyclic intermediate gives a 5,7-bicyclic skeleton with a more stable exocyclic tertiary carbocation. Finally, incorporation of a chloride anion from  $\text{TiCl}_4$  or solvent followed by protonolysis of the titanate provides the 5,7-bicyclic compound **6**.

#### Transannular cyclization of **3** and **5** leading to the compounds having a 6,7-bicyclic skeleton

We examined the transannular cyclization of zerumbone derivatives such as **3** and **5**. We envisioned transannular cyclization between C-4 and C-8 of **3b** with a divalent palladium complex<sup>16</sup> leading to a compound having a 5,8-bicyclic sesquiterpene skeleton, *i.e.* asteriscane. The reaction of **3b** with a stoichiometric amount of  $\text{Pd}(\text{OAc})_2$  in MeCN at  $50\text{ }^{\circ}\text{C}$ ,



however, gave **7b** with the bicyclo[5.4.0]undec-8-ene structure in 47% yield [eqn. (5)]. This result shows that an expected oxy- $\pi$ -allylpalladium species undergoes the carbopalladation in 6-*endo* fashion rather than 5-*exo* fashion to give a 6,7-bicyclic alkylpalladium intermediate leading to **7b** (Scheme 5). The *cis*



Scheme 5

stereochemistry at the ring juncture of **7b** was deduced by NOE analysis.<sup>17</sup>

Next, we investigated the transannular reaction of **5**. We supposed that the minor product **4a** shown in eqn. (3) would be produced by formation of **5a**, and then we tried to find an

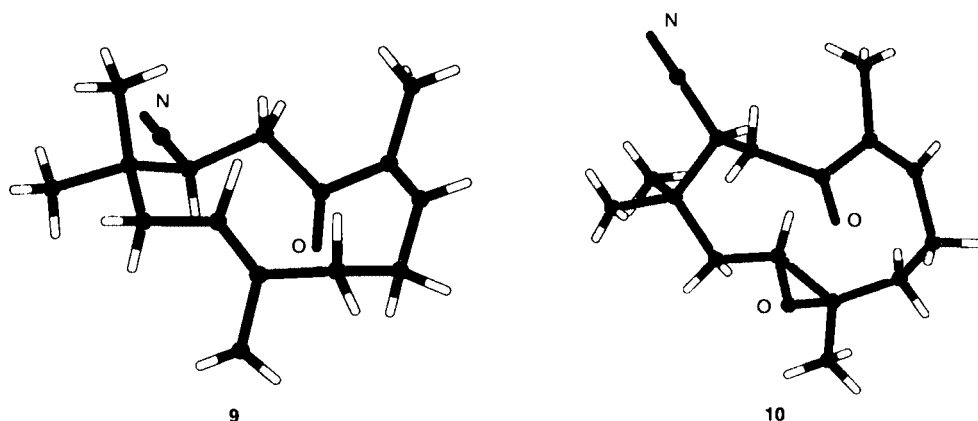
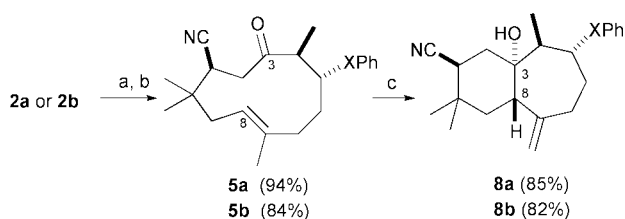


Fig. 3 Molecular structures of **9** and **10**.

effective route for desilylation of **3** leading to **5**. The reaction of **2a** and **2b** with TMS-CN followed by treatment with HF·pyridine complex selectively gave **5a** and **5b**, in 94 and 84% yields for the two steps, respectively (Scheme 6). As expected, **5a** and **5b**

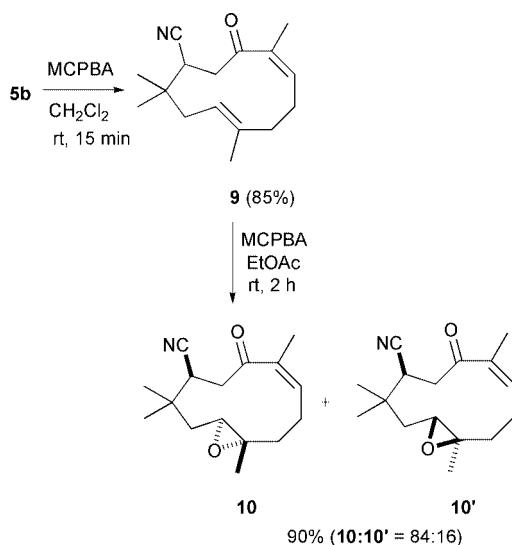


Scheme 6 Reagents and conditions: (a) TMS-CN,  $\text{BF}_3 \cdot \text{OEt}_2$ ,  $\text{CHCl}_3$ , reflux, 0.5 h; (b) HF·py, MeCN, rt, 2 h; (c) HCl aq.,  $\text{CHCl}_3$ , rt, 1.5 h.

were converted to compounds with a 6,7-bicyclic skeleton, **8a** and **8b**, by the acid-catalyzed transannular reaction between C-3 and C-8 of **5**. All spectral data supported the structure of **8** having an *exo* methylene, which is similar to that of **4a**.

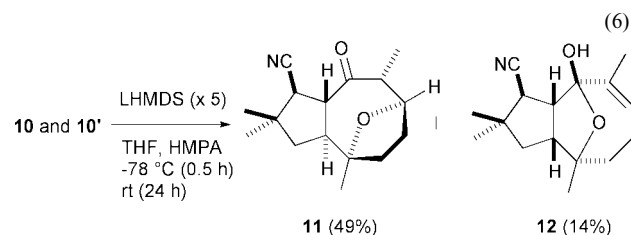
#### Transannular cyclization leading to compounds with a 5,8-bicyclic skeleton

As mentioned above the palladium-promoted transannular reaction of **3b** was found to produce a compound with a 6,7-bicyclic and not a 5,8-bicyclic skeleton. During the course of our study on the transformation of zerumbone derivatives, we found that (1*Z*)-5-cyanozerumbone **9** was obtained by selenoxide elimination of **5b** (Scheme 7). This selenoxide elimination was almost regioselective with only traces of (8*E*,11*E*)-5-cyano-2,6,6,9-tetramethylcycloundeca-8,11-dien-3-one (unconjugated



Scheme 7

product) being obtained. This stereochemical outcome strongly supported the *syn*-vicinity of hydrogen and PhSe at C-1 and C-2 of **5b**. The X-ray structure of **4a** (Fig. 1) also supported the orientation and stereochemistry of the elimination (*vide supra*). The structure of **9** was also established by X-ray crystallographic analysis (Fig. 3); the X-ray structure of **9** clearly shows that the two double bonds  $\Delta^{1,2}$  and  $\Delta^{8,9}$  adopt a parallel arrangement in the solid-state structure. Then, we carried out epoxidation of the  $\Delta^{8,9}$  double bond of **9** expecting stereoselective epoxidation through the *Re*-face of the  $\Delta^{8,9}$  double bond leading to the (8*R*\*,9*R*\*)-epoxide. Upon treatment of **9** with MCPBA in AcOEt, a mixture of 8,9-epoxides **10** and **10'** was produced in 90% yield in a ratio of 84:16 (Scheme 7). The structure of the major product **10** was unambiguously determined by X-ray crystallographic analysis as shown in Fig. 3. This compound contained the opposite epoxide configuration (8*S*\*,9*S*\*) to the expected one (8*R*\*,9*R*\*). Anticipating the transannular displacement of an epoxide by an enolate, as demonstrated in fusicoccadione from dolabelladione monoepoxide by Shin and Fenical,<sup>18</sup> we examined the bond-forming reaction between C-4 and C-8 of **10** and **10'** in order to obtain compounds with a 5,8-bicyclic skeleton. When the reaction of a mixture of **10** and **10'** with lithium



hexamethyldisilazane (LHMDS) was carried out in THF at  $-78^\circ\text{C}$ , tricyclic products **11** and **12** were obtained in 49 and 14% yields, respectively [eqn. (6)]. The structures of these products were characterized by spectroscopic data and unambiguously determined by X-ray diffraction analysis (Fig. 4). The major product **11** was found to involve a 5,8-bicyclic skeleton in *trans* angular configuration, while the minor product **12** was found to involve a 5,8-bicyclic hemiketal structure with a *cis* angular configuration. Although the *trans* angular configuration in the major product we obtained is rare in natural sesquiterpenes, this result shows that the transannular reaction between C-4 and C-8 of zerumbone derivatives can provide the 5,8-bicyclic structure. The stereochemical configuration at C-9 of each product indicates that **11** and **12** might be produced from **10** and **10'**, respectively (Scheme 8). Therefore, we can expect the predominant formation of the 5,8-bicyclic structure with a *cis* angular configuration, provided that we can obtain the reverse stereoselectivity in the epoxidation of **9**.

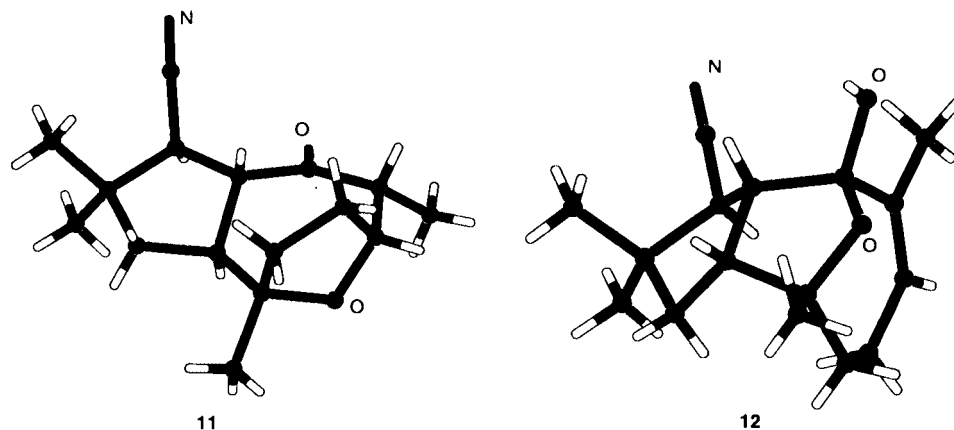
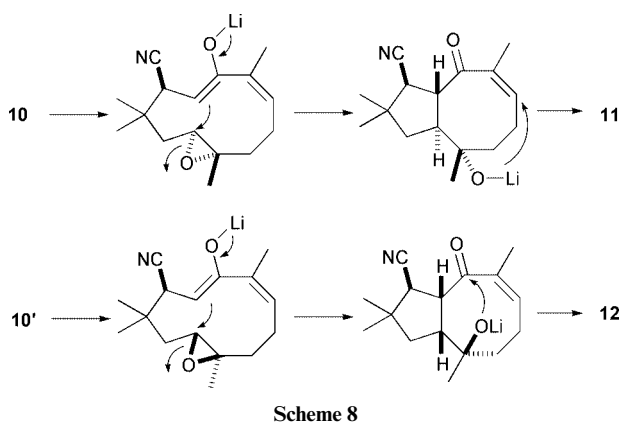


Fig. 4 Molecular structures of **11** and **12**.



Scheme 8

In conclusion, we have demonstrated that the stepwise chemical transformation of zerumbone using conjugate addition reactions of its penta-1,4-dien-3-one moiety afforded zerumbone derivatives bearing functionalities, which can be applied to transannular cyclization reactions that lead to more complex polycyclic sesquiterpene structures. Transannular cyclization of **3** or **5** induced by palladium or a Brønsted acid gives 6,7-bicyclic structures *via* bond formation between C-4 and C-9, or between C-3 and C-8. The Lewis acid-promoted transannular reaction of **2** followed by rearrangement produces a compound with a bicyclo[5.3.0]-decane structure. Intramolecular nucleophilic ring-opening of an epoxide with a lithium enolate is followed by transannular cyclization to give the novel polycyclic structures **11** and **12**.

## Experimental

### General

All reactions were performed in oven-dried or flame-dried glassware under an atmosphere of dry Ar or N<sub>2</sub> unless otherwise noted. Solvents and chemicals were obtained commercially and purified by standard procedures. <sup>1</sup>H NMR spectra were recorded on 270 or 400 MHz FT-NMR spectrometers, and <sup>13</sup>C NMR spectra were recorded on 67.5 or 100 MHz FT-NMR spectrometers. Chemical shifts are reported in ppm relative to TMS in the solvents specified. <sup>1</sup>H NMR data are reported as follows: chemical shift in ppm ( $\delta$ ), multiplicity (s=singlet, d=doublet, t=triplet, q=quartet, sept=septet, m=multiplet), coupling constant (Hz), relative intensity, and interpretation. <sup>13</sup>C NMR data are reported as follows: chemical shift in ppm ( $\delta$ ). Melting points are uncorrected. Analytical thin layer chromatography (TLC) was performed with silica gel 60 Merck F-254 plates. Column chromatography on SiO<sub>2</sub> was performed with Merck silica gel 60. Elemental analyses were performed at Microanalytical Center of Kyoto University.

### Typical procedure for conjugate addition of PhXH (X = S, Se) to **1**

**(4E,8E)-2,6,6,9-Tetramethyl-1-phenylthiocycloundeca-4,8-dien-3-one (2a)**. To a solution of **1** (1.1 g, 5.0 mmol) and benzenethiol (0.56 g, 5.5 mmol) in dry tetrahydrofuran (THF) (5 ml) was added a solution of 1 M tetra-*n*-butylammonium fluoride (TBAF) in THF solution (0.5 ml, 0.5 mmol) at 0 °C under nitrogen. The reaction mixture was stirred at room temperature for 2 h, poured into saturated aqueous NaHCO<sub>3</sub> solution (50 ml), and extracted with AcOEt (3 × 20 ml). The organic layer was dried over MgSO<sub>4</sub>. The solvent was evaporated under vacuum and the residual solid was recrystallized from hexane–AcOEt (20:1) to afford a pure colorless crystalline solid **2a** (1.1 g, 3.4 mmol, 67% yield); mp 122.8–124.1 °C;  $\nu_{\max}$  (KBr)/cm<sup>-1</sup> 1686 (C=O), 1622 (C=C);  $\delta_{\text{H}}$  (400 MHz, CDCl<sub>3</sub>) 1.12 (s, 3H), 1.14 (s, 3H), 1.28 (d, *J* 6.8 Hz, 3H), 1.43 (m, 1H), 1.53 (s, 3H), 1.82 (m, 1H), 1.89–1.96 (m, 2H), 2.10 (dd, *J* 6.0, 13.0 Hz, 1H), 2.29 (t, *J* 12.7 Hz, 1H), 2.57 (dq, *J* 6.8, 6.8 Hz, 1H), 3.30 (dt, *J* 4.0, 8.4 Hz, 1H), 5.10 (dd, *J* 4.8, 11.8 Hz, 1H), 5.78 (d, *J* 16.6 Hz, 1H), 6.17 (d, *J* 16.6 Hz, 1H), 7.23–7.32 (m, 3H), 7.42–7.46 (m, 2H);  $\delta_{\text{C}}$  (100 MHz, CDCl<sub>3</sub>) 13.5, 16.7, 23.5, 29.1, 33.5, 38.8, 39.3, 41.6, 52.6, 54.1, 123.3, 124.9, 127.5, 128.9, 133.3, 135.2, 137.2, 153.5, 204.2. C<sub>21</sub>H<sub>28</sub>OS requires C, 76.78; H, 8.59. Found: C, 76.52; H, 8.73%.

**(4E,8E)-2,6,6,9-Tetramethyl-1-phenylselenocycloundeca-4,8-dien-3-one (2b)**. A crystalline colorless solid (65% yield); mp 109.8–111.5 °C;  $\nu_{\max}$  (KBr)/cm<sup>-1</sup> 1686 (C=O), 1628 (C=C);  $\delta_{\text{H}}$  (400 MHz, CDCl<sub>3</sub>) 1.08 (s, 3H), 1.14 (s, 3H), 1.28 (d, *J* 6.8 Hz, 3H), 1.44–1.58 (m, 4H), 1.84–2.00 (m, 3H), 2.08 (br dd, *J* 10.0, 13.2 Hz, 1H), 2.28 (dd, *J* 7.2, 12.8 Hz, 1H), 2.55 (dq, *J* 6.8, 7.8 Hz, 1H), 3.25–3.34 (m, 1H), 5.05 (br dd, *J* 4.4, 11.6 Hz, 1H), 5.75 (d, *J* 16.0 Hz, 1H), 6.18 (d, *J* 16.0 Hz, 1H), 7.23–7.33 (m, 3H), 7.54–7.61 (m, 2H);  $\delta_{\text{C}}$  (100 MHz, CDCl<sub>3</sub>) 15.2, 16.8, 23.4, 29.1, 38.9, 39.3, 41.6, 49.6, 54.0, 123.3, 124.8, 127.9, 129.0, 129.2, 135.7, 137.1, 153.6, 204.7. C<sub>21</sub>H<sub>28</sub>OSe requires C, 67.19; H, 7.52. Found: C, 67.16; H, 7.44%.

### 5-Cyano-2,6,6,9-tetramethyl-1-phenylseleno-3-(trimethylsilyloxy)cycloundeca-3,8-diene (**3b**)

To a solution of **2b** (0.38 g, 1.0 mmol) and trimethylsilyl cyanide (0.40 ml, 3.0 mmol) in dry CHCl<sub>3</sub> (0.5 ml) was added boron trifluoride–diethyl ether complex (BF<sub>3</sub>·OEt<sub>2</sub>) (13  $\mu$ l, 0.10 mmol) at reflux under nitrogen and the mixture was stirred for 0.5 h. After the mixture was diluted with CHCl<sub>3</sub> (10 ml), the solution was washed with water (10 ml) and then by saturated aqueous NaHCO<sub>3</sub> solution (10 ml) and dried over MgSO<sub>4</sub>. The solvent was evaporated under vacuum and the residue was subjected to short column chromatography on SiO<sub>2</sub> with hexane–AcOEt (10:1) as an eluent to give a colorless liquid of **3b** as a mixture

of two diastereoisomers (dr = 91:9) (0.31 g, 0.64 mmol, 64% yield);  $\nu_{\max}$  (neat)/ $\text{cm}^{-1}$  2240 (CN), 1661 (C=C), 1580 (C=C);  $\delta_{\text{H}}$  (400 MHz,  $\text{CDCl}_3$ ) 0.21 (s, 9H), 1.07 (s, 3H), 1.21–1.23 (m, 6H), 1.56 (s, 3H), 1.72–2.08 (m, 7H), 2.68 (ddd,  $J$  2.0, 6.0, 11.2 Hz, 1H), 3.41 (d,  $J$  10.4 Hz, 1H), 4.46 (d,  $J$  10.4 Hz, 1H), 5.11 (br dd,  $J$  7.2, 8.0 Hz, 1H), 7.23–7.32 (m, 3H), 7.51–7.60 (m, 2H);  $\delta_{\text{C}}$  (100 MHz,  $\text{CDCl}_3$ ) 0.7, 18.1, 21.3, 21.9, 31.4, 32.7, 37.2, 37.4, 38.4, 41.7, 47.3, 55.9, 103.0, 120.4, 120.8, 127.6, 129.0, 130.1, 133.8, 135.0, 158.0.  $\text{C}_{25}\text{H}_{37}\text{NOSeSi}$  requires C, 63.27; H, 7.86; N, 2.95. Found: C, 63.02; H, 7.86; N, 2.84%.

#### 10-Cyano-6-fluoro-2,6,9,9-tetramethyl-3-phenylthiobicyclo[5.4.0]undecan-1-ol (4a)

To a solution of **2a** (0.66 g, 2.0 mmol) and trimethylsilyl cyanide (0.80 ml, 6.0 mmol) in dry  $\text{CHCl}_3$  (1.0 ml) was added boron trifluoride–diethyl ether complex ( $\text{BF}_3 \cdot \text{OEt}_2$ ) (26  $\mu\text{l}$ , 0.20 mmol) under nitrogen. The reaction mixture was stirred at reflux temperature for 0.5 h. This solution was washed with water (20 ml) and extracted with  $\text{Et}_2\text{O}$  ( $2 \times 20$  ml). The combined organic layer was dried over  $\text{Na}_2\text{SO}_4$  and then evaporated under vacuum to leave the crude phenylthio analog of **3b** (**3a**). It was added to 47% aqueous HF (1.0 ml) in MeCN (50 ml) at room temperature. The mixture was stirred for 4 h. The resulting mixture was poured into saturated aqueous  $\text{NaHCO}_3$  solution (50 ml) and extracted with AcOEt ( $2 \times 50$  ml). The organic layer was dried over  $\text{MgSO}_4$ . The solvent was evaporated under vacuum and the residue was subjected to column chromatography on  $\text{SiO}_2$  with hexane–AcOEt (3:1) to give a colorless solid **4a** (0.13 g, 0.36 mmol, 18% yield); mp 64.5–67.3 °C;  $\nu_{\max}$  (KBr)/ $\text{cm}^{-1}$  3452 (OH), 2241 (CN);  $\delta_{\text{H}}$  (400 MHz,  $\text{CDCl}_3$ ) 1.00 (s, 3H), 1.17 (s, 3H), 1.21 (d,  $J$  8.0 Hz, 3H), 1.35 (s, 3H), 1.46 (dd,  $J$  13.8, 3.8 Hz, 1H), 1.61–1.64 (m, 2H), 1.72–1.77 (m, 2H), 1.89–2.10 (m, 3H), 2.19–2.25 (m, 1H), 2.18 (q,  $J$  8.0 Hz, 1H), 2.56 (br s, 1H), 2.82 (dd,  $J$  12.8, 3.6 Hz, 1H), 3.42 (t,  $J$  9.4 Hz, 1H), 7.26–7.30 (m, 2H), 7.32–7.39 (m, 3H);  $\delta_{\text{C}}$  (100 MHz,  $\text{CDCl}_3$ ) 19.7, 21.4, 26.2, 27.7, 30.7, 32.5, 35.6, 36.5, 41.2, 42.1, 44.5, 47.6, 51.0, 75.6, 76.2, 121.4, 127.5, 129.3, 131.7, 134.5.  $\text{C}_{22}\text{H}_{30}\text{FNOS}$  requires C, 70.36; H, 8.05; N, 3.73. Found: C, 70.46; H, 8.33; N, 3.46%.

#### Typical procedure for conjugate addition of TMSCN to 2

(*8E*)-5-Cyano-2,6,6,9-tetramethyl-1-phenylthiocycloundec-8-en-3-one (**5a**). To a solution of **2a** (0.99 g, 3.0 mmol) and trimethylsilyl cyanide (1.2 ml, 9.0 mmol) in dry  $\text{CHCl}_3$  (3.0 ml) was added  $\text{BF}_3 \cdot \text{Et}_2\text{O}$  (38  $\mu\text{l}$ , 0.30 mmol) at reflux under nitrogen. After the mixture was stirred for 0.5 h, the solution was added to a solution of hydrogen fluoride–pyridine (0.1 ml) in MeCN (40 ml) at room temperature. (CAUTION: HCN gas is generated.) After stirring for 2 h, the solution was washed with saturated aqueous  $\text{NaHCO}_3$  solution (50 ml). The aqueous layer was extracted with AcOEt ( $2 \times 20$  ml). The combined organic layer was washed with brine (50 ml) and dried over  $\text{MgSO}_4$ . The solvent was evaporated under vacuum and the residue was subjected to column chromatography on  $\text{SiO}_2$  with hexane–AcOEt (10:1) to give a white solid of **5a** (1.0 g, 2.8 mmol, 94% yield); mp 83.8–86.0 °C;  $\nu_{\max}$  (neat)/ $\text{cm}^{-1}$  2231 (CN), 1698 (C=O);  $\delta_{\text{H}}$  (400 MHz,  $\text{CDCl}_3$ ) 1.13 (s, 3H), 1.26 (s, 3H), 1.28 (d,  $J$  6.8 Hz, 3H), 1.49 (s, 3H), 1.55–1.77 (m, 1H), 1.94–2.13 (m, 4H), 2.13–2.24 (m, 1H), 2.48 (dd,  $J$  6.0, 17.6 Hz, 1H), 2.80–2.98 (m, 3H), 3.07–3.18 (m, 1H), 5.18 (t,  $J$  10.8 Hz, 1H), 7.21–7.35 (m, 3H), 7.37–7.46 (m, 2H);  $\delta_{\text{C}}$  (100 MHz,  $\text{CDCl}_3$ ) 15.2, 16.5, 24.0, 30.0, 30.1, 34.7, 36.6, 36.9, 37.8, 39.6, 48.6, 52.2, 121.3, 121.8, 127.5, 129.1, 132.8, 134.6, 137.4, 206.6.  $\text{C}_{22}\text{H}_{29}\text{NOS}$  requires C, 74.32; H, 8.22; N, 3.94. Found: C, 74.05; H, 8.34; N, 3.71%.

(*8E*)-5-Cyano-2,6,6,9-tetramethyl-1-phenylselenocycloundec-8-en-3-one (**5b**). A colorless liquid (84% yield);  $\nu_{\max}$  (neat)/ $\text{cm}^{-1}$  2240 (CN), 1712 (C=O);  $\delta_{\text{H}}$  (400 MHz,  $\text{CDCl}_3$ ) 1.11 (s, 3H),

1.21–1.28 (m, 6H), 1.46 (s, 3H), 1.73–1.86 (m, 1H), 1.90–2.20 (m, 5H), 2.43 (dd,  $J$  7.6, 17.6 Hz, 1H), 2.80–2.97 (m, 3H), 3.07–3.20 (m, 1H), 5.15 (br t,  $J$  7.2 Hz, 1H), 7.23–7.36 (m, 3H), 7.51–7.60 (m, 2H);  $\delta_{\text{C}}$  (100 MHz,  $\text{CDCl}_3$ ) 15.8, 16.3, 23.7, 29.9, 31.2, 34.6, 36.4, 37.3, 37.4, 39.6, 48.4, 48.9, 121.1, 121.7, 128.0, 128.7, 129.0, 135.3, 137.0, 206.7.  $\text{C}_{22}\text{H}_{29}\text{NOSe}$  requires C, 65.66; H, 7.26; N, 3.48. Found: C, 65.51; H, 7.41; N, 3.41%.

#### Typical procedure for $\text{TiCl}_4$ -promoted transannular cyclization of 2

10-(1-Chloro-1-methylethyl)-3,7-dimethyl-4-phenylthiobicyclo[5.3.0]decan-2-one (**6a**). To a solution of **2a** (0.26 g, 0.80 mmol) in  $\text{CH}_2\text{Cl}_2$  (10 ml) was slowly added  $\text{TiCl}_4$  (0.16 ml, 1.8 mmol) at  $-40$  °C under nitrogen. The reaction mixture was stirred at  $-40$  °C for 2 h. The resulting mixture was washed with saturated aqueous  $\text{NaHCO}_3$  solution (30 ml) and the aqueous layer was extracted with  $\text{CH}_2\text{Cl}_2$  ( $2 \times 30$  ml). The combined organic layers were dried over  $\text{MgSO}_4$ . The solvent was evaporated under vacuum and the residue was subjected to column chromatography on  $\text{SiO}_2$  with hexane–AcOEt (20:1) to give a white solid of **6a** (0.17 g, 0.46 mmol, 57% yield); mp 88.9–90.4 °C;  $\nu_{\max}$  (KBr)/ $\text{cm}^{-1}$  1693 (C=O);  $\delta_{\text{H}}$  (400 MHz,  $\text{CDCl}_3$ ) 0.71 (s, 3H), 1.39 (s, 3H), 1.49 (d,  $J$  7.6 Hz, 3H), 1.50–1.71 (m, 4H), 1.55 (s, 3H), 1.76–1.82 (m, 2H), 1.89–2.00 (m, 1H), 2.13 (dd,  $J$  4.8, 14.4 Hz, 1H), 2.41 (dq,  $J$  7.2, 10.8 Hz, 1H), 2.95 (m, 1H), 3.11 (d,  $J$  11.2 Hz, 1H), 3.18 (t,  $J$  10.4 Hz, 1H), 7.24–7.34 (m, 3H), 7.40–7.43 (m, 2H);  $\delta_{\text{C}}$  (100 MHz,  $\text{CDCl}_3$ ) 20.4, 20.5, 25.7, 32.2, 32.3, 32.6, 43.2, 43.3, 46.6, 50.0, 52.3, 54.2, 59.3, 75.6, 127.2, 129.0, 132.2, 134.3, 212.3.  $\text{C}_{21}\text{H}_{29}\text{ClOS}$  requires C, 69.11; H, 8.01. Found: C, 69.33; H, 8.29%.

10-(1-Chloro-1-methylethyl)-3,7-dimethyl-4-phenylselenobicyclo[5.3.0]decan-2-one (**6b**). A white solid (59% yield); mp 74.6–78.0 °C;  $\nu_{\max}$  (KBr)/ $\text{cm}^{-1}$  1699 (C=O);  $\delta_{\text{H}}$  (400 MHz,  $\text{CDCl}_3$ ) 0.68 (s, 3H), 1.37 (s, 3H), 1.47 (d,  $J$  7.2 Hz, 1H), 1.48–1.81 (m, 6H), 1.54 (s, 3H), 1.91 (m, 1H), 2.26 (m, 1H), 2.44 (dq,  $J$  7.2, 11.2 Hz, 1H), 2.94 (ddd,  $J$  4.4, 7.2, 10.8 Hz, 1H), 3.10 (d,  $J$  8.8 Hz, 1H), 3.24 (t,  $J$  10.8 Hz, 1H), 7.27–7.33 (m, 3H), 7.55–7.58 (m, 2H);  $\delta_{\text{C}}$  (100 MHz,  $\text{CDCl}_3$ ) 20.5, 21.6, 25.7, 32.2, 32.3, 33.6, 43.3, 44.0, 46.4, 48.2, 50.0, 55.1, 59.2, 75.6, 127.9, 128.4, 129.1, 135.3, 212.5.  $\text{C}_{21}\text{H}_{29}\text{ClOSe}$  requires C, 61.24; H, 7.10. Found: C, 61.51; H, 7.31%.

#### 11-Cyano-3,7,10,10-tetramethyl-4-phenylselenobicyclo[5.4.0]undec-8-en-2-one (**7b**)

To a solution of **3b** (0.10 g, 0.20 mmol) in MeCN (2.0 ml) was added  $\text{Pd}(\text{OAc})_2$  (0.045 g, 0.20 mmol) at 50 °C under nitrogen. After stirring for 10 h, the solution was filtered through a Celite pad and the filtrate was washed with water (10 ml). The organic layer was dried over  $\text{MgSO}_4$ . The solvent was evaporated under vacuum and the residue was subjected to column chromatography on  $\text{SiO}_2$  with hexane–AcOEt (10:1) to give a colorless oil of **7b** (0.038 g, 0.095 mmol, 47% yield);  $\nu_{\max}$  (neat)/ $\text{cm}^{-1}$  2243 (CN), 1707 (C=O);  $\delta_{\text{H}}$  (400 MHz,  $\text{CDCl}_3$ ) 0.74 (s, 3H), 1.18 (s, 3H), 1.20 (s, 3H), 1.48 (d,  $J$  6.8 Hz, 3H), 1.55–1.77 (m, 3H), 2.30 (br dd,  $J$  4.0, 11.2 Hz, 1H), 2.54 (dq,  $J$  6.8, 11.2 Hz, 1H), 2.89 (d,  $J$  12.0 Hz, 1H), 3.06 (dd,  $J$  11.2, 11.2 Hz, 1H), 3.14 (d,  $J$  12.0 Hz, 1H), 5.19 (d,  $J$  9.6 Hz, 1H), 5.32 (d,  $J$  9.6 Hz, 1H), 7.25–7.39 (m, 3H), 7.53–7.60 (m, 2H);  $\delta_{\text{C}}$  (100 MHz,  $\text{CDCl}_3$ ) 18.8, 20.9, 25.3, 29.1, 33.2, 34.1, 37.0, 37.3, 44.7, 47.3, 50.8, 54.9, 120.0, 127.9, 128.2, 129.2, 133.8, 135.6, 211.2.  $\text{C}_{22}\text{H}_{27}\text{NOSe}$  requires C, 65.99; H, 6.80; N, 3.50. Found: C, 65.89; H, 6.95; N, 3.34%.

#### 10-Cyano-1-hydroxy-2,9,9-trimethyl-6-methylene-3-phenylthiobicyclo[5.4.0]undecane (**8a**)

To a solution of **5a** (0.071 g, 0.20 mmol) in  $\text{CHCl}_3$  (1 ml) was

added conc. HCl (one drop) at room temperature. After stirring for 1.5 h, the solution was washed with saturated aqueous NaHCO<sub>3</sub> solution (5 ml) and the aqueous layer was extracted with CHCl<sub>3</sub> (3 × 10 ml). The combined organic layers were dried over MgSO<sub>4</sub>. The solvent was evaporated under vacuum and the residue was subjected to column chromatography on SiO<sub>2</sub> with hexane–AcOEt (20:1) to give a white solid of **8a** (0.063 g, 0.17 mmol, 85%); mp 91.4–93.1 °C;  $\nu_{\max}$  (KBr)/cm<sup>-1</sup> 3505 (OH), 2243 (CN);  $\delta_{\text{H}}$  (400 MHz, CDCl<sub>3</sub>) 1.05–1.11 (m, 6H), 1.15 (s, 3H), 1.25 (dd, *J* 3.2, 13.2 Hz, 1H), 1.79 (dd, *J* 4.0, 14.0 Hz, 1H), 1.88 (t, *J* 13.2 Hz, 2H), 1.96–2.11 (m, 2H), 2.11–2.23 (m, 1H), 2.23–2.33 (m, 1H), 2.42 (dd, *J* 3.2, 13.2 Hz, 1H), 2.60–2.71 (m, 1H), 2.87 (dd, *J* 4.0, 12.8 Hz, 1H), 3.30–3.45 (m, 2H), 4.88 (s, 1H), 5.09 (s, 1H), 7.22–7.36 (m, 3H), 7.38–7.45 (m, 2H);  $\delta_{\text{C}}$  (100 MHz, CDCl<sub>3</sub>) 16.9, 20.4, 30.0, 30.5, 32.4, 34.5, 35.0, 36.4, 41.0, 41.1, 46.6, 53.3, 74.3, 114.7, 121.6, 127.4, 129.1, 132.0, 135.5, 149.4. C<sub>22</sub>H<sub>29</sub>NOS requires C, 74.32; H, 8.22; N, 3.94. Found: C, 74.07; H, 8.38; N, 3.92%.

#### 10-Cyano-1-hydroxy-2,9,9-trimethyl-6-methylene-3-phenyl-selenobicyclo[5.4.0]undecane (**8b**)

A white solid (82% yield); mp 117.4–118.8 °C;  $\nu_{\max}$  (KBr)/cm<sup>-1</sup> 3457 (OH), 2236 (CN);  $\delta_{\text{H}}$  (400 MHz, CDCl<sub>3</sub>) 1.09 (s, 3H), 1.10 (d, *J* 8.0 Hz, 3H), 1.16 (s, 3H), 1.25 (dd, *J* 2.8, 13.2 Hz, 1H), 1.78–1.94 (m, 3H), 2.02–2.19 (m, 2H), 2.42 (dd, *J* 3.2, 13.2 Hz, 1H), 2.53–2.66 (m, 1H), 2.86 (dd, *J* 4.0, 12.0 Hz, 1H), 3.13 (br s, 1H), 3.39 (br t, *J* 7.2 Hz, 1H), 4.91 (s, 1H), 5.12 (s, 1H), 7.25–7.34 (m, 3H), 7.51–7.60 (m, 2H);  $\delta_{\text{C}}$  (100 MHz, CDCl<sub>3</sub>) 17.7, 20.4, 30.5, 30.5, 32.4, 35.0, 35.8, 36.1, 40.8, 41.0, 47.9, 49.4, 74.2, 115.2, 121.5, 127.8, 129.2, 131.1, 134.3, 149.1. C<sub>22</sub>H<sub>29</sub>NOSe requires C, 65.66; H, 7.26; N, 3.48. Found: C, 65.53; H, 7.33; N, 3.46%.

#### (1Z,8E)-5-Cyano-2,6,6,9-tetramethylcycloundeca-1,8-dien-3-one (**9**)

To a solution of **5b** (0.96 g, 2.4 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (20 ml) was added *m*-chloroperbenzoic acid (MCPBA) (0.56 g, 2.6 mmol) at 0 °C. The solution was stirred at room temperature for 0.5 h. The solution was washed with saturated aqueous Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> solution (30 ml) and saturated aqueous Na<sub>2</sub>CO<sub>3</sub> solution (30 ml). The organic layer was dried over MgSO<sub>4</sub>. The solvent was evaporated under vacuum and the residue was subjected to column chromatography on SiO<sub>2</sub> with hexane–AcOEt (8:1) to give a white solid of **9** (0.50 g, 2.0 mmol, 85% yield); mp 79.4–81.1 °C;  $\nu_{\max}$  (KBr)/cm<sup>-1</sup> 2246 (CN), 1694 (C=O), 1635 (C=C);  $\delta_{\text{H}}$  (400 MHz, CDCl<sub>3</sub>) 1.12 (s, 3H), 1.25 (s, 3H), 1.49 (s, 3H), 1.86 (br ddd, *J* 2.8, 10.0, 12.0 Hz, 1H), 1.91–2.10 (m, 6H), 2.13–2.26 (m, 1H), 2.32–2.47 (m, 1H), 2.57 (dd, *J* 4.4, 18.0 Hz, 1H), 2.68 (dd, *J* 4.4, 18.0 Hz, 1H), 3.13 (t, *J* 4.4 Hz, 1H), 4.89 (br dd, *J* 3.2, 9.2 Hz, 1H), 5.65 (br dt, *J* 1.2, 8.8 Hz, 1H);  $\delta_{\text{C}}$  (100 MHz, CDCl<sub>3</sub>) 17.1, 20.5, 24.4, 24.4, 28.5, 34.9, 35.6, 38.1, 38.5, 40.9, 120.3, 122.4, 134.0, 137.2, 138.1, 202.1. C<sub>16</sub>H<sub>23</sub>NO requires C, 78.32; H, 9.45; N, 5.71. Found: C, 78.11; H, 9.44; N, 5.55%.

#### (1Z)-5-Cyano-2,6,6,9-tetramethyl-8,9-epoxycycloundec-1-en-3-one (**10/10'**)

To a solution of **9** (0.75 g, 3.1 mmol) in AcOEt (30 ml) was added MCPBA (0.72 g, 3.4 mmol) at 0 °C and the resulting solution was stirred at room temperature for 2 h. It was washed with saturated aqueous Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> solution (30 ml) and saturated aqueous Na<sub>2</sub>CO<sub>3</sub> solution (30 ml). The organic layer was separated and dried over MgSO<sub>4</sub>. The solvent was evaporated under vacuum and the residue was subjected to column chromatography on SiO<sub>2</sub> with hexane–AcOEt (4:1) to give a white solid of **10** (0.72 g, 2.8 mmol, 90%, **10/10'** = 84:16). Pure **10** was obtained by recrystallization from hexane–AcOEt

(10:1); mp 95.3–97.8 °C;  $\nu_{\max}$  (KBr)/cm<sup>-1</sup> 2243 (CN), 1686 (C=O), 1623 (C=C);  $\delta_{\text{H}}$  (400 MHz, CDCl<sub>3</sub>) 1.15 (s, 3H), 1.27 (s, 3H), 1.28–1.34 (br m, 4H), 1.45–1.49 (m, 1H), 1.68–1.82 (m, 1H), 1.99 (s, 3H), 2.09–2.21 (m, 2H), 2.60 (dd, *J* 2.8, 6.0 Hz, 1H), 2.64–2.89 (m, 3H), 3.07 (dd, *J* 5.2, 6.8 Hz, 1H), 5.82 (t, *J* 8.0 Hz, 1H);  $\delta_{\text{C}}$  (100 MHz, CDCl<sub>3</sub>) 17.8, 20.7, 24.5, 26.2, 26.8, 36.2, 36.6, 36.9, 37.0, 41.2, 56.6, 60.5, 120.6, 136.5, 137.5, 203.2. C<sub>16</sub>H<sub>23</sub>NO<sub>2</sub> requires C, 73.53; H, 8.87; N, 5.36. Found: C, 73.32; H, 8.92; N, 5.26%.

#### Tricyclic ketone **11**

To a solution of **10/10'** (0.13 g, 0.50 mmol, dr = 84:16) in THF (5.0 ml) was added lithium hexamethyldisilazane (LHMDS, 1 M in THF, 0.60 ml, 0.60 mmol) at –78 °C and the mixture was stirred at –78 °C for 0.5 h under nitrogen. The solution was stirred at room temperature for 2 h. The solution was poured into saturated aqueous NH<sub>4</sub>Cl solution (30 ml) and extracted with Et<sub>2</sub>O (3 × 20 ml). The organic layer was dried over MgSO<sub>4</sub>. The solvent was evaporated under vacuum and the residue was subjected to column chromatography on SiO<sub>2</sub> with hexane–AcOEt (6:1) to give two tricyclic compounds of **11** (0.05 g, 0.2 mmol, 40%) and **12** (0.02 g, 0.08 mmol, 15%). A white solid **11**; mp 121.4–122.8 °C;  $\nu_{\max}$  (KBr)/cm<sup>-1</sup> 2240 (CN), 1694 (C=O);  $\delta_{\text{H}}$  (400 MHz, CDCl<sub>3</sub>) 1.18 (s, 3H), 1.20 (s, 3H), 1.22 (s, 3H), 1.24 (s, 3H), 1.28–1.35 (m, 1H), 1.64–1.76 (m, 2H), 1.88 (dt, *J* 8.4, 11.6 Hz, 1H), 2.07–2.20 (m, 2H), 2.41 (q, *J* 7.6 Hz, 1H), 2.63 (m, 1H), 3.07 (d, *J* 10.8 Hz, 1H), 3.19 (t, *J* 11.6 Hz, 1H), 4.35 (dd, *J* 4.0, 9.6 Hz, 1H);  $\delta_{\text{C}}$  (100 MHz, CDCl<sub>3</sub>) 16.3, 26.6, 28.9, 30.3, 31.4, 34.1, 38.4, 41.2, 43.6, 53.4, 54.4, 58.1, 80.6, 85.1, 119.6, 206.9. C<sub>16</sub>H<sub>23</sub>NO<sub>2</sub> requires C, 73.53; H, 8.87; N, 5.36. Found: C, 73.30; H, 8.96; N, 5.23%.

**Tricyclic hemiketal 12.** A white solid (15% yield); mp 146.8–148.0 °C;  $\nu_{\max}$  (KBr)/cm<sup>-1</sup> 3356 (OH), 2234 (CN);  $\delta_{\text{H}}$  (400 MHz, CDCl<sub>3</sub>) 1.18 (s, 3H), 1.24 (s, 3H), 1.28 (s, 3H), 1.43–1.83 (m, 4H), 1.87 (s, 3H), 1.89–1.96 (m, 1H), 2.41–2.49 (m, 1H), 2.56 (d, *J* 11.6 Hz, 1H), 2.90 (br s, 1H), 2.99–3.06 (m, 1H), 3.18 (t, *J* 11.6 Hz, 1H), 5.80 (dd, *J* 1.2, 8.0 Hz, 1H);  $\delta_{\text{C}}$  (100 MHz, CDCl<sub>3</sub>) 20.1, 23.2, 25.8, 27.3, 28.5, 36.7, 38.3, 42.8, 49.3, 53.8, 61.6, 81.9, 108.1, 120.4, 130.5, 137.6. C<sub>16</sub>H<sub>23</sub>NO<sub>2</sub> requires C, 73.53; H, 8.87; N, 5.36. Found: C, 73.53; H, 8.64; N, 5.25%.

#### X-Ray structure determinations †

Measurements were made on a Rigaku AFC-7R four-circle automated diffractometer with Mo-K $\alpha$  radiation ( $\lambda = 0.71069$  Å). Structures were solved by Patterson Methods (DIRDIF92 PATTY)<sup>19</sup> or direct methods (SIR92)<sup>20</sup> and expanded using Fourier techniques.

Crystal data for C<sub>22</sub>H<sub>30</sub>FNOS (**4a**), *M* = 375.54, monoclinic, space group *P*<sub>2</sub><sub>1</sub>/*a* (no. 14), *a* = 11.860(5), *b* = 13.058(3), *c* = 13.784(3) Å,  $\beta = 98.57(2)^\circ$ , *V* = 2110(1) Å<sup>3</sup>, *Z* = 4, *D*<sub>c</sub> = 1.182 g cm<sup>-3</sup>, *F*(000) = 808,  $\mu(\text{Mo-K}\alpha) = 1.72$  cm<sup>-1</sup>. At 296 K, 5302 reflections were measured, 5066 were unique (*R*<sub>int</sub> = 0.048). The final residuals were *R* = 0.047 and *R*<sub>w</sub> = 0.062. The maximum and minimum peaks in the final difference map were 0.44 and –0.37 e Å<sup>-3</sup> respectively.

Crystal data for C<sub>21</sub>H<sub>26</sub>ClOS (**6a**), *M* = 364.97, triclinic, space group *P* $\bar{1}$  (no. 2), *a* = 11.943(7), *b* = 12.196(8), *c* = 7.739(2) Å,  $\alpha = 104.76(3)^\circ$ ,  $\beta = 99.19(3)^\circ$ ,  $\gamma = 69.53(4)^\circ$ , *V* = 1018.1(9) Å<sup>3</sup>, *Z* = 2, *D*<sub>c</sub> = 1.190 g cm<sup>-3</sup>, *F*(000) = 392,  $\mu(\text{Mo-K}\alpha) = 2.95$  cm<sup>-1</sup>. At 296 K, 4901 reflections were measured, 4677 were unique (*R*<sub>int</sub> = 0.012). The final residuals were *R* = 0.061 and *R*<sub>w</sub> = 0.064. The maximum and minimum peaks in the final difference map were 0.24 and –0.35 e Å<sup>-3</sup> respectively.

† CCDC reference number 207/484. See <http://www.rsc.org/suppdata/pl/b0/b004284f/> for crystallographic files in .cif format.

Crystal data for C<sub>16</sub>H<sub>23</sub>NO (**9**),  $M = 245.36$ , monoclinic, space group  $P2_1/n$  (no. 14),  $a = 23.302(4)$ ,  $b = 6.339(4)$ ,  $c = 13.214(3)$  Å,  $\beta = 105.64(2)^\circ$ ,  $V = 1440.0(9)$  Å<sup>3</sup>,  $Z = 4$ ,  $D_c = 1.132$  g cm<sup>-3</sup>,  $F(000) = 536$ ,  $\mu(\text{Mo-K}\alpha) = 0.69$  cm<sup>-1</sup>. At 296 K, 3395 reflections were measured, 3305 were unique ( $R_{\text{int}} = 0.024$ ). The final residuals were  $R = 0.041$  and  $R_w = 0.037$ . The maximum and minimum peaks in the final difference map were 0.16 and  $-0.11$  e Å<sup>-3</sup> respectively.

Crystal data for C<sub>16</sub>H<sub>23</sub>NO<sub>2</sub> (**10**),  $M = 261.36$ , monoclinic, space group  $P2_1/n$  (no. 14),  $a = 16.025(6)$ ,  $b = 5.876(4)$ ,  $c = 17.157(6)$  Å,  $\beta = 112.60(2)^\circ$ ,  $V = 1491(1)$  Å<sup>3</sup>,  $Z = 4$ ,  $D_c = 1.164$  g cm<sup>-3</sup>,  $F(000) = 568$ ,  $\mu(\text{Mo-K}\alpha) = 0.76$  cm<sup>-1</sup>. At 296 K, 3542 reflections were measured, 3422 were unique ( $R_{\text{int}} = 0.029$ ). The final residuals were  $R = 0.040$  and  $R_w = 0.034$ . The maximum and minimum peaks in the final difference map were 0.17 and  $-0.13$  e Å<sup>-3</sup> respectively.

Crystal data for C<sub>64</sub>H<sub>92</sub>N<sub>4</sub>O<sub>8</sub> (**11**),  $M = 1045.45$ , triclinic, space group  $P\bar{1}$  (no. 2),  $a = 8.090(4)$ ,  $b = 39.650(8)$ ,  $c = 9.371(2)$  Å,  $\alpha = 90.00(2)^\circ$ ,  $\beta = 89.99(3)^\circ$ ,  $\gamma = 90.17(3)^\circ$ ,  $V = 3006(1)$  Å<sup>3</sup>,  $Z = 8$ ,  $D_c = 4.620$  g cm<sup>-3</sup>,  $F(000) = 4544$ ,  $\mu(\text{Mo-K}\alpha) = 3.01$  cm<sup>-1</sup>. At 296 K, 7377 reflections were measured, 6886 were unique ( $R_{\text{int}} = 0.035$ ). The final residuals were  $R = 0.042$  and  $R_w = 0.041$ . The maximum and minimum peaks in the final difference map were 0.09 and  $-0.12$  e Å<sup>-3</sup> respectively.

Crystal data for C<sub>16</sub>H<sub>23</sub>NO<sub>2</sub> (**12**),  $M = 261.36$ , monoclinic, space group  $P2_1/n$  (no. 14),  $a = 13.385(3)$ ,  $b = 16.622(3)$ ,  $c = 13.449(3)$  Å,  $\beta = 92.15(2)^\circ$ ,  $V = 2990.1(9)$  Å<sup>3</sup>,  $Z = 8$ ,  $D_c = 1.161$  g cm<sup>-3</sup>,  $F(000) = 1136$ ,  $\mu(\text{Mo-K}\alpha) = 0.76$  cm<sup>-1</sup>. At 296 K, 7162 reflections were measured, 6876 were unique ( $R_{\text{int}} = 0.013$ ). The final residuals were  $R = 0.051$  and  $R_w = 0.047$ . The maximum and minimum peaks in the final difference map were 0.54 and  $-0.20$  e Å<sup>-3</sup> respectively.

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